

This article was downloaded by:

On: 22 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713454007>

Chemical constituents of the brown alga *Dictyopteris divaricata*

F. -H. Song; X. Fan; X. -L. Xu; J. -L. Zhao; L. -J. Han; J. -G. Shi

To cite this Article Song, F. -H. , Fan, X. , Xu, X. -L. , Zhao, J. -L. , Han, L. -J. and Shi, J. -G.(2005) 'Chemical constituents of the brown alga *Dictyopteris divaricata*', Journal of Asian Natural Products Research, 7: 6, 777 — 781

To link to this Article: DOI: 10.1080/1028602032000169532

URL: <http://dx.doi.org/10.1080/1028602032000169532>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Chemical constituents of the brown alga *Dictyopteris divaricata*

F.-H. SONG^{†‡}, X. FAN[†], X.-L. XU^{†‡}, J.-L. ZHAO[¶], L.-J. HAN[†] and J.-G. SHI^{¶*}

[†]Institute of Oceanology, Chinese Academy of Sciences, Qingdao 266071, China

[‡]Graduate School of the Chinese Academy of Sciences, Beijing 100039, China

[¶]Institute of Materia Medica, Chinese Academy of Medical Sciences and Peking Union Medical College, Beijing 100050, China

(Received 3 September 2003; revised 25 September 2003; in final form 29 September 2003)

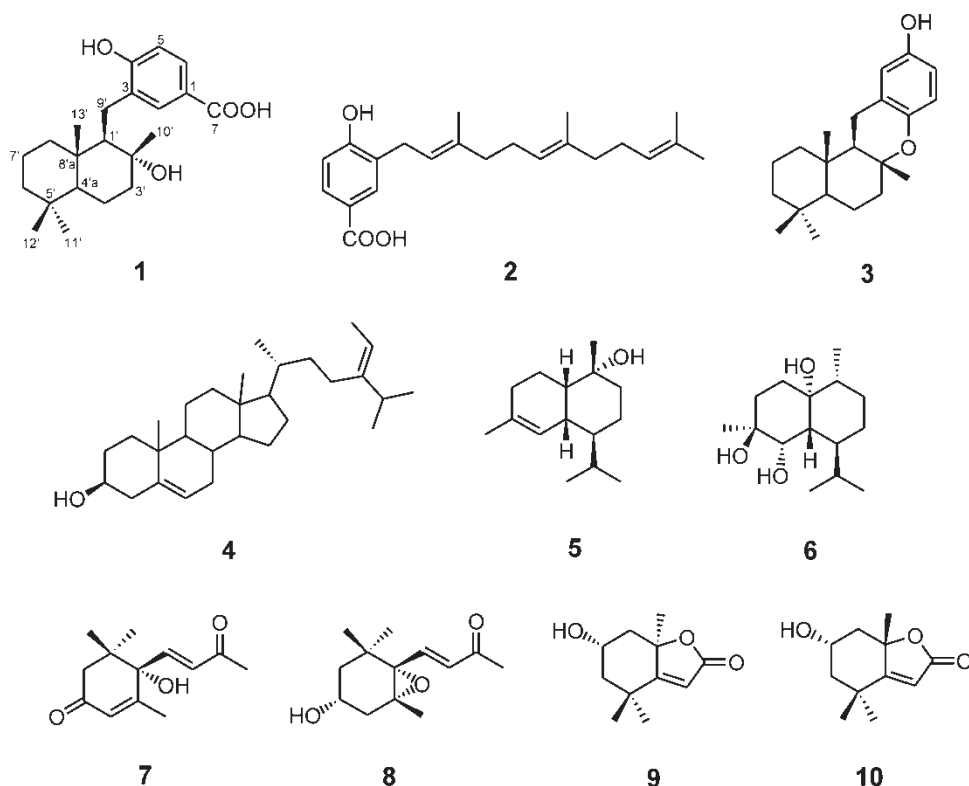
A novel sesquiterpene-substituted benzoic acid, named dictyvaric acid (**1**), together with nine known compounds (**2**–**10**), have been isolated from the brown alga *Dictyopteris divaricata* Okam. The structure of **1** was elucidated as 3-[(decahydro-2-hydroxy-2,5,5,8a-tetramethyl-1-naphthalenyl)-methyl]-4-hydroxybenzoic acid by spectroscopic methods, including IR, FABMS, HR-FABMS, 1D and 2D NMR techniques. All compounds were obtained from this species for the first time.

Keywords: Brown alga; *Dictyopteris divaricata* Okam; Sesquiterpene-substituted benzoic acid; Dictyvaric acid; 3-[(Decahydro-2-hydroxy-2,5,5,8a-tetramethyl-1-naphthalenyl)-methyl]-4-hydroxybenzoic acid

1. Introduction

Dictyopteris divaricata Okam. is a brown alga belonging to the Dictyotaceae family, and is widely distributed along the coast of the Yellow Sea, China. Although various sesquiterpene-substituted phenols [1–4], sesquiterpenes [5–10] and C₁₁ hydrocarbons [11–18] have been isolated from species of *Dictyopteris*, few compounds have been reported from the title species [7–9]. In our investigation of the chemical diversity of Chinese seaweeds, a new sesquiterpene-substituted phenol, named dictyvaric acid (**1**) and the nine known compounds 3-farnesyl-*p*-hydroxybenzoic acid (**2**) [19], chromazonarol (**3**) [2,20], fucosterol (**4**) [21], (–)-torreyol (**5**) [22], 4β,5α-dihydroxycubanol (**6**) [10], dehydrovomifoliol (**7**) [23], 3β-hydroxy-5α,6α-epoxy-7-megastigmen-9-one (**8**) [24], loliolide (**9**) [25] and isolololide (**10**) [26], have been isolated from the ethanolic extract of *D. divaricata* Okam. The structures of the known compounds were determined on the basis of comparing their spectral data with those of reported in the literature. This paper reports the structure elucidation of the new compound (**1**).

*Corresponding author. E-mail: shijg@imm.ac.cn



2. Results and discussion

The EtOAc-soluble fraction of the ethanolic extract of the brown alga *Dictyopteria divaricata* Okam. was chromatographed successively over silica gel, Sephadex LH-20 and by reversed-phase HPLC to afford compounds (1–10). The structures of the known compounds (2–10) were identified by comparison of their spectral data with those reported in literature.

Compound **1** (white needles) was assigned a molecular formula of $C_{22}H_{32}O_4$ by negative-ion HR-FABMS at m/z 359.2248 ($[M - H]^-$). Its IR spectrum (KBr) exhibits a strong, broad absorption band for hydroxyl groups (3429 cm^{-1}), and bands for conjugated carbonyl (1678 cm^{-1}) and an aromatic ring (1614 , 1587 and 1502 cm^{-1}). The ^1H NMR spectrum of **1** shows an ABX system at δ 7.63 (1H, dd, $J = 10.5, 2.0\text{ Hz}$), 6.68 (1H, d, $J = 10.5\text{ Hz}$) and 7.78 (1H, d, $J = 2.0\text{ Hz}$), which is attributed to a 1,3,4-trisubstituted benzene moiety, and signals assignable to four methyl groups attached to quaternary carbons at δ 0.77 (3H, s), 0.80 (3H, s), 0.94 (3H, s) and 1.21 (3H, s). The ^{13}C NMR and DEPT spectra of **1** showed 22 carbon signals, including one carboxyl, three quaternary carbons (one oxygenated), two methines and six methylenes besides the signals attributed to four methyl groups and the 1,3,4-trisubstituted benzene moiety (see table 1). These data suggest that compound **1** has a sesquiterpene-substituted phenol structure [2]. A comparison of the ^1H and ^{13}C NMR data of **1** with those of sesquiterpene-substituted phenol yahazunol [2] demonstrated that the 1,4-dihydroxy-phenyl moiety in yahazunol was replaced by the 3-substituted 4-hydroxybenzoic

Table 1. ^1H and ^{13}C NMR data of compound **1**^a (δ ppm, J Hz).

No.	δ_{H}	δ_{C}	No.	δ_{H}	δ_{C}
1		123.1s	5'		34.1s
2	7.78 (d, 1H, 2.0)	134.5d	6' α	1.01 (td, 1H, 13.0, 3.5)	
3		131.1s	6' β	1.26 (m, 1H)	43.0t
4		161.0s	7' α	1.55 (m, 1H)	
5	6.68 (d, 1H, 10.5)	116.7d	7' β	1.27 (m, 1H)	19.3t
6	7.63 (dd, 1H, 10.5, 2.0)	130.2d	8' α	0.57 (td, 1H, 13.0, 3.5)	
7		171.7s	8' β	1.69 (br.d, 1H, 13.0)	41.7t
1'	1.53 (dd, 1H, 11.5, 1.5)	62.9d	8'a		41.0s
2'		75.3s	9' α	2.56 (dd, 1H, 15.0, 11.5)	
3' α	1.48 (td, 1H, 13.0, 3.5)	44.7t	9' β	2.80 (dd, 1H, 15.0, 1.5)	27.3t
3' β	1.84 (dt, 1H, 13.0, 3.5)		10'	1.21 (s, 3H)	24.2q
4' α	1.63 (ddt, 1H, 13.0, 2.0, 3.5)	21.4t	11'	0.80 (s, 3H)	33.9q
4' β	1.30 (ddt, 1H, 13.0, 2.0, 3.5)		12'	0.77 (s, 3H)	21.9q
4'a	0.89 (dd, 1H, 13.0, 2.0)	57.5d	13'	0.94 (s, 3H)	15.9q

^aNMR data measured in CD_3OD at 500 MHz for proton and at 125 MHz for carbon; assignments based on DEPT, ^1H - ^1H COSY, HMQC and HMBC experiments.

acid moiety in **1**. The structure of **1** was further confirmed by 2D NMR experiments. The proton signals in the ^1H NMR spectrum and carbon signals in ^{13}C NMR spectrum were unambiguously assigned by ^1H - ^1H COSY and HMQC experiments (see table 1). In the HMBC spectrum (see figure 1), correlations from H-9' to C-1', C-2' and C-8a', H-10' to C-1', C-2' and C-3', both H-11' and H-12' to C-4a', C-5' and C-6', H-13' to C-1', C-8' and C-8a' confirm the sesquiterpene moiety, while correlations from both H-2 and H-6 to C-4 and C-7, and from H-5 to C-1 and C-3 confirm the 3-substituted-4-hydroxybenzoic acid moiety. In addition, the connectivity between the two moieties was established by the correlations from H-9' to C-2, C-3 and C-4, and from H-2 to C-9'. The NOESY spectrum of **1** exhibits NOE correlations among H-10', H-12' and H-13', and between H-1' and H-4a', indicating that **1** and yahazunol possess identical relative stereochemistry (figure 2). Therefore, the structure of **1** was determined as 3-[(decahydro-2-hydroxy-2,5,5,8a-tetramethyl-1-naphthalenyl)-methyl]-4-hydroxybenzoic acid, named dictyvaric acid.

3. Experimental section

3.1 General experimental procedures

Melting points were determined on an XT-4 micro melting point apparatus and are uncorrected. Optical rotations were measured with Perkin-Elmer 241 polarimeter.

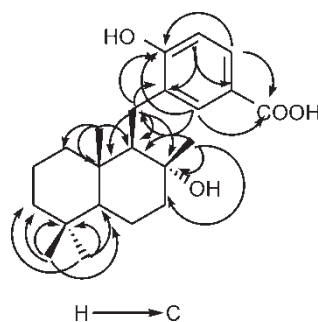
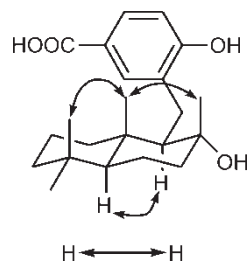


Figure 1. Key HMBC correlations of **1**.

Figure 2. Key NOESY correlations of **1**.

IR spectra were recorded as KBr disks on a Nicolet Impact 400 FT-IR spectrophotometer. NMR spectra were obtained at 500 and 125 MHz for ^1H and ^{13}C , respectively, on an Inova 500 MHz spectrometer in CD_3OD with solvent peaks as references. The negative-ion FABMS and HR-FABMS were obtained on a Micromass Autospec-Ultima ETOF spectrometer. Column chromatography was performed with silica gel (200–300 mesh) and Sephadex LH-20. HPLC was performed using an Alltima C18 preparative column (10 μ , 22 \times 250 mm). TLC was carried out with glass precoated silica gel GF254 plates. Spots were visualized under UV light or by spraying with 5% sulfuric acid in EtOH followed by heating.

3.2 Plant material

The brown alga *Dictyopteris divaricata* Okam. was collected at the coast of Qingdao, Shandong, China, in April 2002 and was identified by Dr Kui-Shuang Shao (Institute of Oceanology, Chinese Academy of Sciences).

3.2.1 Extraction and isolation. The brown alga *Dictyopteris divaricata* Okam. (4.40 kg) was extracted with 95% EtOH at room temperature for 3 \times 72 h, and the solvent was removed under reduced pressure at $<40^\circ\text{C}$ to give a dark residue (320.2 g). The residue was suspended in water and then partitioned with EtOAc. The EtOAc-soluble fraction (120.3 g) was chromatographed over silica gel, eluting with an increasing gradient of Me_2CO (0–100%) in light petroleum. The first fraction, eluted by 20% Me_2CO in light petroleum, was chromatographed over Sephadex LH-20 using light petroleum– CHCl_3 –MeOH (5:5:1) to afford eight subfractions. The seventh subfraction was further separated by reversed-phase preparative HPLC using MeOH as mobile phase to yield compounds **1** (10 mg) and **2** (30 mg). The fifth subfraction was further separated by reversed-phase preparative HPLC using MeOH– H_2O (80:20) as mobile phase to yield compounds **9** (11 mg) and **10** (9 mg).

The second fraction, eluted by 3% Me_2CO in light petroleum, was chromatographed over Sephadex LH-20 using light petroleum– CHCl_3 –MeOH (5:5:1) to afford four subfractions. The second subfraction was further separated by reversed-phase preparative HPLC, using MeOH– H_2O (80:20), as mobile phase to yield compound **3** (8 mg), and the third subfraction was recrystallized in acetone to yield compound **4** (15 mg).

The fraction eluted by 2% Me_2CO in light petroleum was recrystallized in acetone to yield compounds **5** (4 mg) and **6** (18 mg). The fraction eluted by 10% Me_2CO in light petroleum was chromatographed over Sephadex LH-20 using light petroleum– CHCl_3 –MeOH (5:5:1) to afford four subfractions, the fourth of which was further separated by reversed-phase

preparative HPLC, using MeOH–H₂O (80:20) as mobile phase, to yield compounds **7** (8 mg) and **8** (13 mg).

3.3 Dictyvaric acid (1)

White needles (MeOH), mp 256–258°C; $[\alpha]_D^{20} + 29^\circ$ (c 0.1, CHCl₃); IR (KBr) ν_{\max} (cm⁻¹); 3429, 2939, 2679, 1678, 1614, 1587, 1450, 1388, 1298, 1244, 1122, 937, 758, 640. ¹H NMR (CD₃OD, 500 MHz) and ¹³C NMR (CD₃OD, 125 MHz) see table 1; negative-ion FABMS m/z 359 [M – H]⁻; negative-ion HR-FABMS m/z 359.2248 (calcd for C₂₂H₃₁O₄, 359.2222).

Acknowledgements

The authors are grateful to Professor Ablez Zeper for mass spectra measurements. This research was financially supported by the National High Technology Development Project (863 Project) (No: 2001AA620403).

References

- [1] W. Fenical, J.J. Sims, K. Squatrito, R.M. Wing, P. Radlick. *J. Org. Chem.*, **38**, 2383 (1973).
- [2] W. Fenical, O. McConnel. *Experientia*, **31**, 1004 (1975).
- [3] G. Cimino, S. Stefano, W. de, Fenical, Minale, L., J.J. Sims. *Experientia*, **31**, 1250 (1975).
- [4] M. Ochi, H. Kotsuki, K. Muraoka, T. Tokoroyama. *Bull. Chem. Soc. Jpn.*, **52**, 629 (1979).
- [5] W. Fenical, J.J. Sims, R.M. Wing, P.C. Radlick. *Phytochemistry*, **11**, 1161 (1972).
- [6] T. Irie, K. Yamamoto, T. Masamune. *Bull. Chem. Soc. Jpn.*, **30**, 1053 (1964).
- [7] M. Suzuki, N. Kowata, E. Kurosawa. *Bull. Chem. Soc. Jpn.*, **54**, 2366 (1981).
- [8] M. Segawa, K. Yamano, H. Shrahama. *Phytochemistry*, **29**, 973 (1990).
- [9] M. Suzuki, N. Kowata, E. Kurosawa, H. Kobayashi, I. Tanaka. *Chem. Lett.*, 2187 (1990).
- [10] G.M. Koenig, A.D. Wright. *Magn. Reson. Chem.*, **33**, 178 (1995).
- [11] R.E. Moore, J.A. Pettus Jr. *Tetrahedron Lett.*, 4787 (1968).
- [12] J.A. Pettus, Jr, R.E. Moore. *J. Chem. Soc., Chem. Commun.*, 1093 (1970).
- [13] P. Roller, K. Au, R.E. Moore. *J. Chem. Soc., Chem. Commun.*, 503 (1971).
- [14] J.A. Pettus, Jr, R.E. Moore. *J. Am. Chem. Soc.*, **93**, 3087 (1971).
- [15] R.E. Moore, G. Yost. *J. Chem. Soc., Chem. Commun.*, 937 (1973).
- [16] R.E. Moore, J.A. Pettus, Jr, J. Mistysyn. *J. Org. Chem.*, **39**, 2201 (1974).
- [17] K. Yamada, H. Tan, H. Tatematsu, M. Ojika. *Tetrahedron*, **42**, 3775 (1986).
- [18] W. Boland, D.G. Müller. *Tetrahedron Lett.*, **28**, 307 (1987).
- [19] B. Chen, K. Kawazoe, Y. Takaishi, G. Honda, M. Itoh, Y. Takeda, O.K. Kodzhimatov, O. Ashurmetov. *J. Nat. Prod.*, **63**, 362 (2000).
- [20] P. Djura, D.B. Stierle, B. Sullivan, D.J. Faulkner. *J. Org. Chem.*, **45**, 1435 (1980).
- [21] J.H. Sheu, G.H. Wang, P.J. Sung, Y.H. Chiu, C.Y. Duh. *Planta Med.*, **63**, 571 (1997).
- [22] A. Karin, B. Karlson, T. Norin. *Tetrahedron*, **37**, 425 (1981).
- [23] M. Takasugi, M. Anetai, N. Katsui, T. Masamune. *Chem. Lett.*, 245 (1973).
- [24] H. Duan, Y. Takaishi, H. Momota, Y. Ohmoto, T. Taki. *Phytochemistry*, **59**, 85 (2002).
- [25] N. Okada, K. Shirata, M. Niwano, H. Koshino, M. Uramoto. *Phytochemistry*, **37**, 281 (1994).
- [26] J. Kimura, N. Maki. *J. Nat. Prod.*, **65**, 57 (2002).